

# Photocatalytic Effects of TiO<sub>2</sub> Nanoparticles: Is there an Impact on the Ecotoxicity of PAHs?

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## Background

- Titanium dioxide nanoparticles (TiO<sub>2</sub> NPs) are used in a variety of consumer products and expected to increasingly enter the aquatic environment [1].
- TiO<sub>2</sub> NPs catalyze the degradation of PAHs under UV irradiation. They are currently being considered for use in the remediation of PAH-polluted soils and wastewaters [2], although little is known about the effects of potential products of this photo modification process to the aquatic environment.
- Photo oxidation of PAHs can alter bioavailability (water solubility) and intrinsic toxicity of contaminants.

## Materials and methods

**Bioassay:** ISO 10872:2010 - Chronic test of *C. elegans* (Nematoda), endpoints: reproduction, fertility and growth

**Test material:**

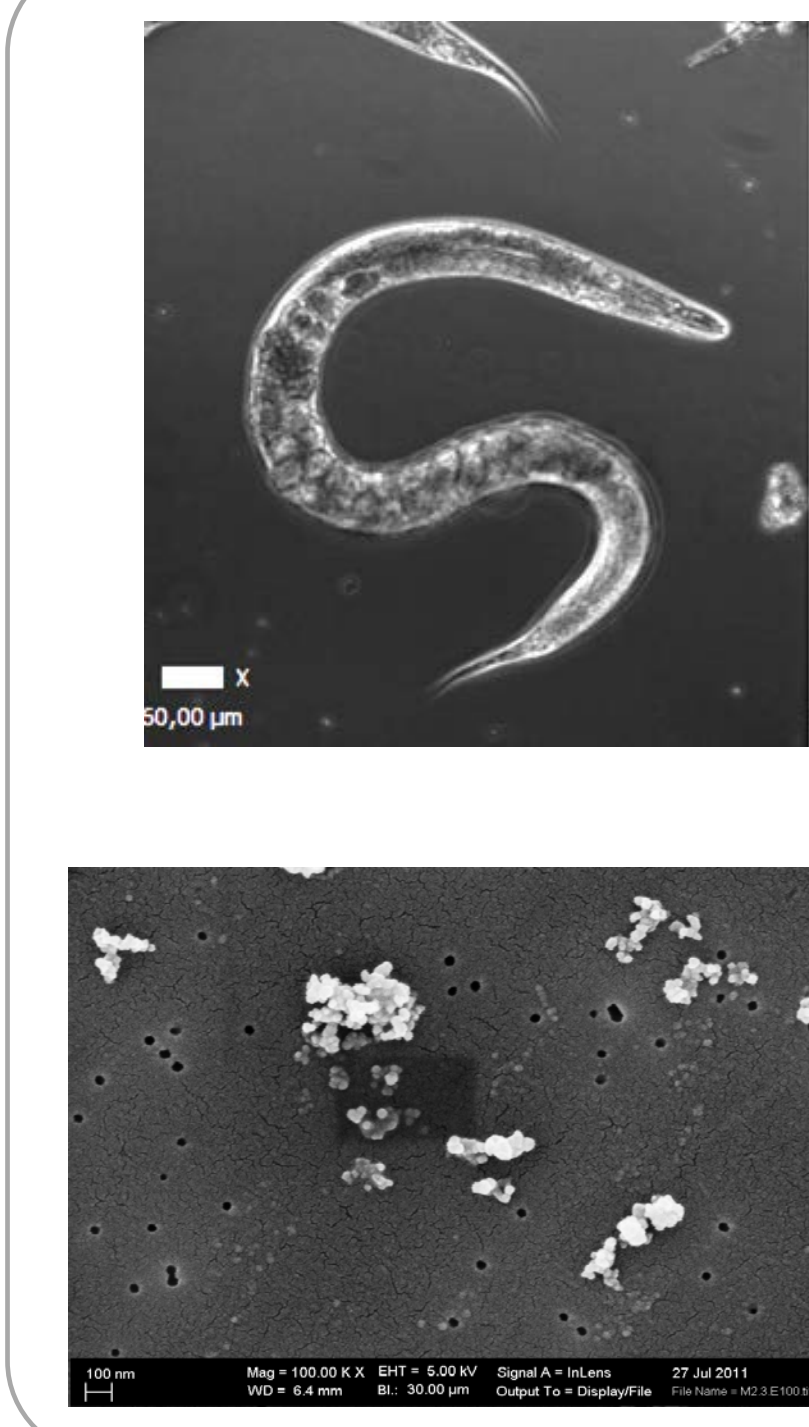
- TiO<sub>2</sub> NPs P 25 Aeroxide (Degussa AG) (primary particle size 21 nm; 75% anatase, 25% rutile)

**Dispersion:** in ultrapure water using magnetic stirring and ultrasonic bath

- Phenanthrene, solvent DMSO

**Particle characterization:**

- Dynamic Light Scattering (DLS) and Zeta-Potential
- Scanning Electron Microscopy (SEM).



**Main objectives:**

- 1) Assess the solar irradiation-mediated ecotoxicological effects of TiO<sub>2</sub> NPs on *Caenorhabditis elegans*
- 2) Investigate the impact of TiO<sub>2</sub> NPs on the toxicity of phenanthrene

## Results

### Effects of solar radiation on TiO<sub>2</sub> toxicity

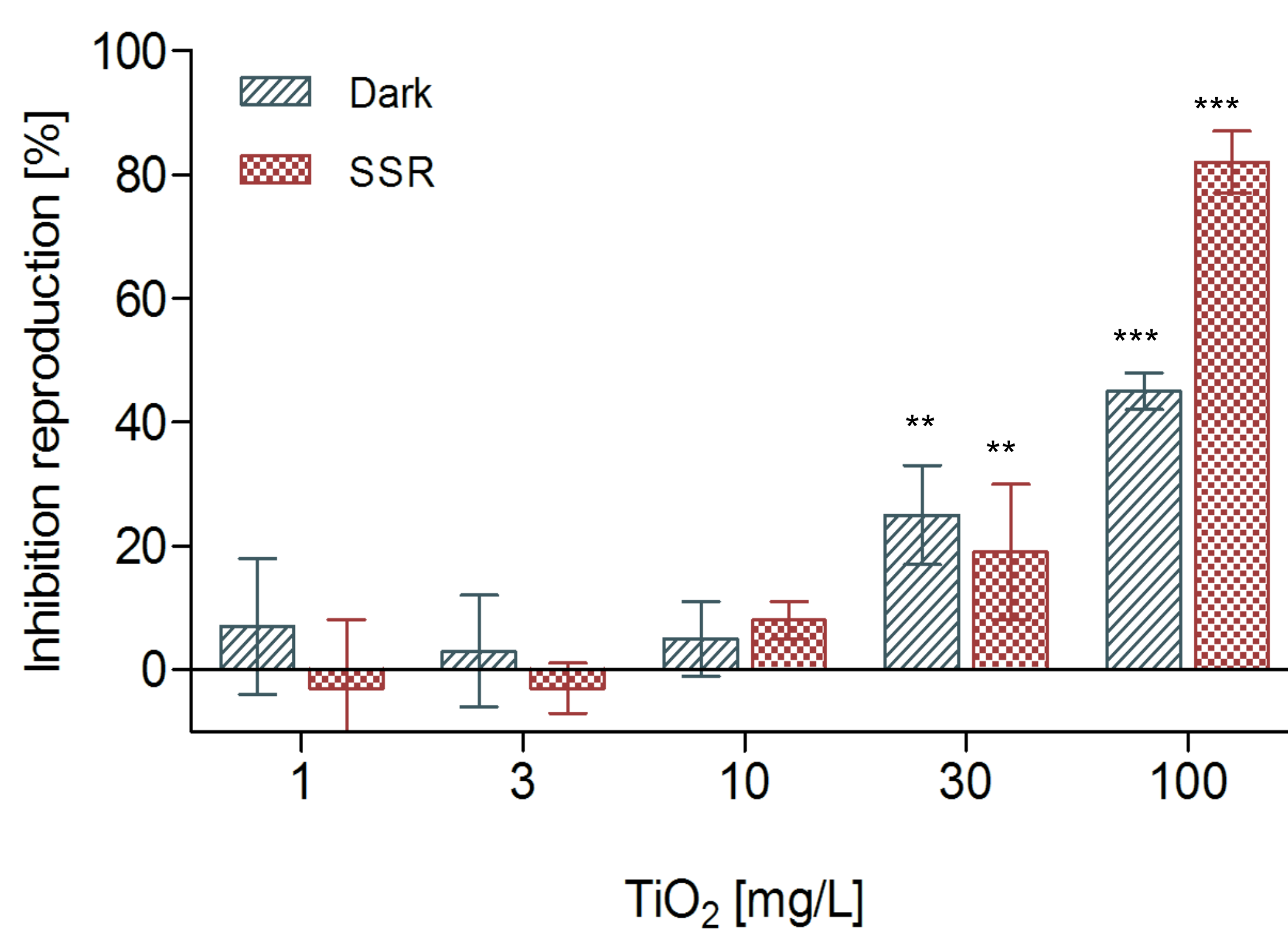


Fig. 1: Inhibition of reproduction of *C. elegans* exposed to TiO<sub>2</sub> with and without SSR [mean inhibition and SD in %, n=4]; p-Values \*\*<0.01 \*\*\* < 0,001 (One-Way ANOVA at  $\alpha = 0.05$ , post test Dunnett). SSR: 30 min irradiation, 4 h after start of the test, 231 W/m<sup>2</sup> at 300-800 nm.

- TiO<sub>2</sub> NPs significantly inhibit the reproduction of *C. elegans* (LOEC 10 mg/L to 30 mg/L)
- 30 min simulated solar radiation (SSR) increases the toxicity by a factor of approx. 2 (EC<sub>50</sub> of 53 mg/L (SSR))

### Effects solar radiation on phenanthrene toxicity

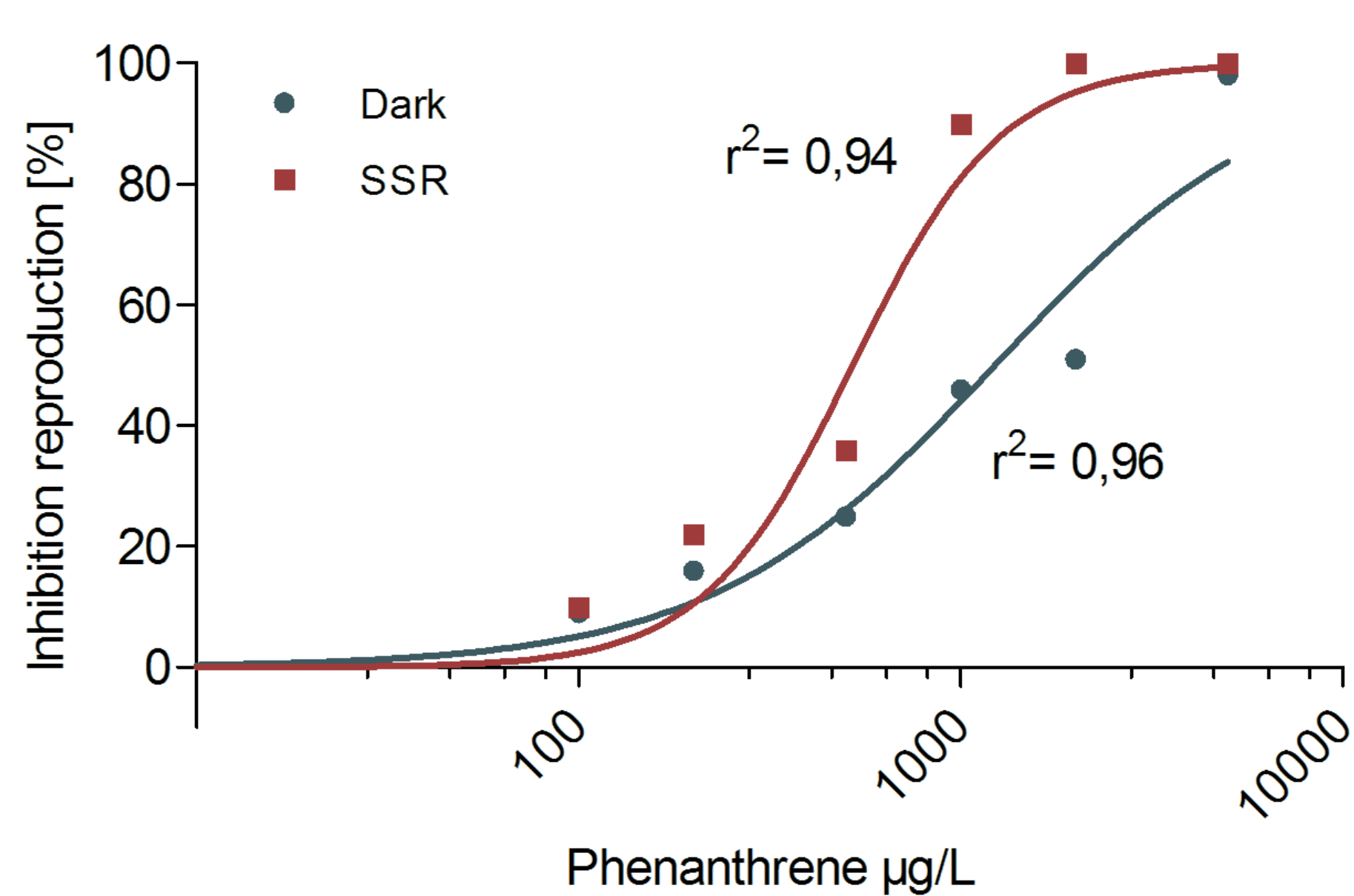


Fig. 2: Inhibition of reproduction of *C. elegans* exposed to phenanthrene with and without SSR [mean inhibition, n=4]; nonlinear fit of log(x). SSR: 30 min irradiation, 4 h after start of the test, 231 W/m<sup>2</sup> at 300-800 nm.

SSR increases the toxicity of phenanthrene by a factor of approx. 2. [EC<sub>50</sub> of 1230  $\mu$ g/L (no-SSR) and 522  $\mu$ g/L (SSR)].

### Effects of solar radiation and TiO<sub>2</sub> on toxicity of phenanthrene (combined exposure)

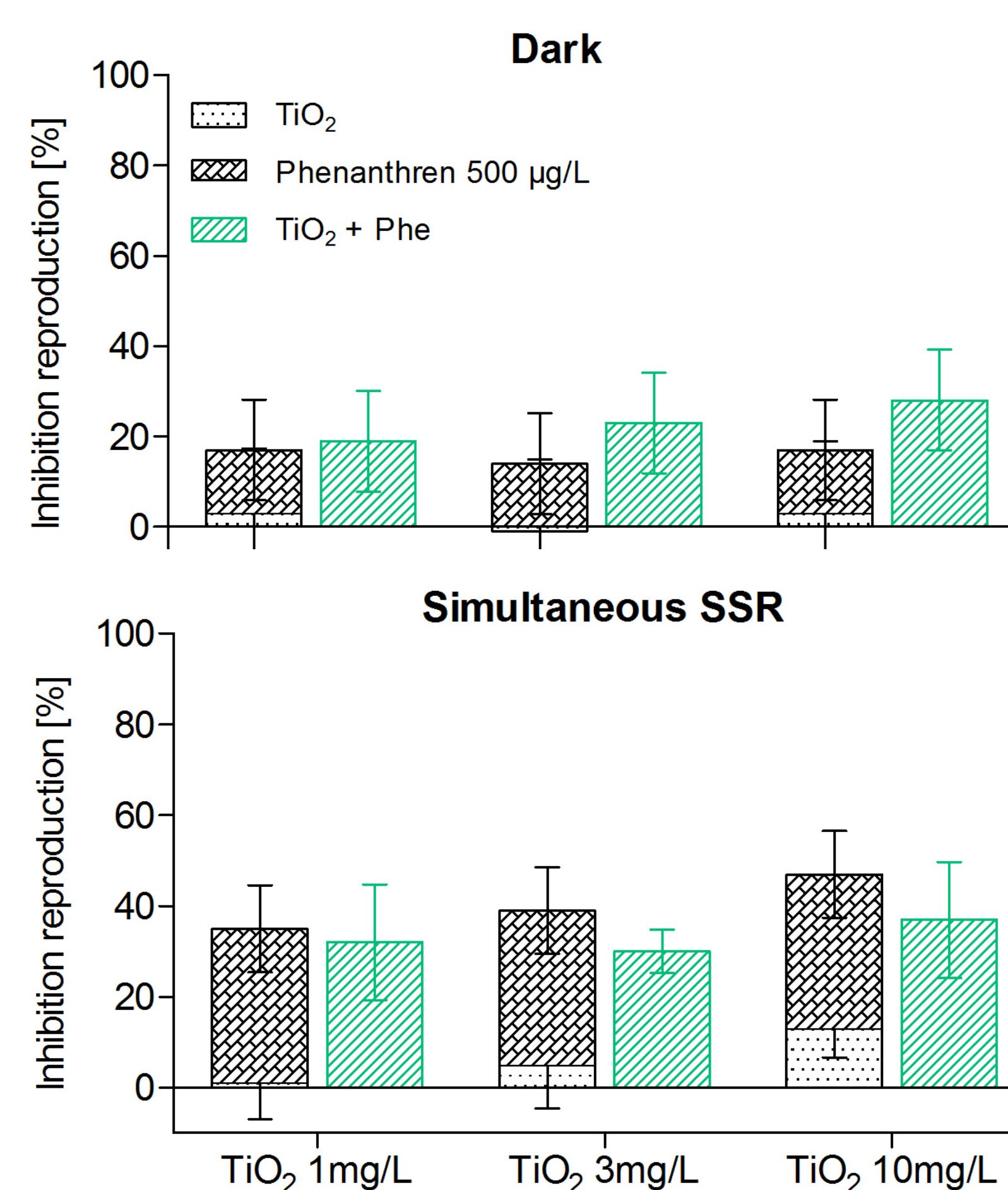


Fig. 3: Inhibition of reproduction of *C. elegans* exposed to phenanthrene and TiO<sub>2</sub> [mean inhibition and SD in %, n=4]. Grey bars: added single substance effects; green bars: effects of combined exposure. SSR: 30 min irradiation, 4 h after test start, 231 W/m<sup>2</sup> at 300-800 nm.

**Dark:** TiO<sub>2</sub> NPs increase the toxicity of phenanthrene slightly, effects are not significant.

**SSR:** Simultaneous irradiation of *C. elegans* exposed to TiO<sub>2</sub> and phenanthrene decreases toxicity slightly, effects are not significant.

## Conclusions

- ❖ Reproduction of *C. elegans* is significantly inhibited by exposure to TiO<sub>2</sub> NPs at concentrations of 10 mg/L and higher
- ❖ Simulated solar radiation increases the inhibition by a factor of approx. 2
- ❖ Phenanthrene toxicity is not significantly affected by simultaneous exposure to TiO<sub>2</sub> or TiO<sub>2</sub> and SSR
- ❖ Agglomeration and sedimentation of the particles is observed in the test system
- ❖ Agglomerated particles are bioavailable for *C. elegans* and ingested into the intestinal tract

### Ingestion of TiO<sub>2</sub> particles

Agglomerated TiO<sub>2</sub> particles are taken up into the worm intestinal tract within 96 h exposure to TiO<sub>2</sub> (Fig. 4 A and B).

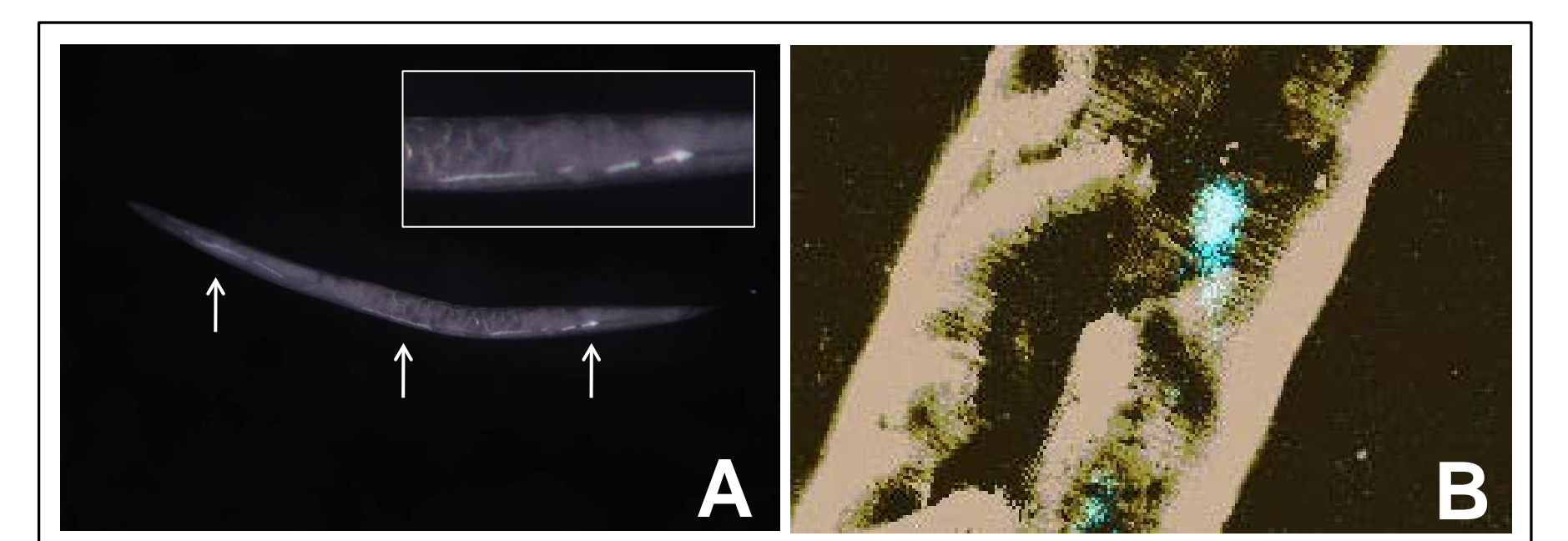


Fig. 4 *C. elegans* exposed to 20mg/L TiO<sub>2</sub> for 96 h; A) Stereomicroscope images B) EDX mapping of Titanium with 20 keV

### Agglomeration of NPs

#### Dynamic Light Scattering

- Mean particle size and SD increase with increasing particle concentrations (253 nm +/-79 to 718 nm +/-145)
- Addition of the test media M9 leads to further agglomeration, particle sizes increase by a factor of 1.15 to 1.92
- Particle sizes do not significantly change during the exposure period of 96 h (Fig. 5)

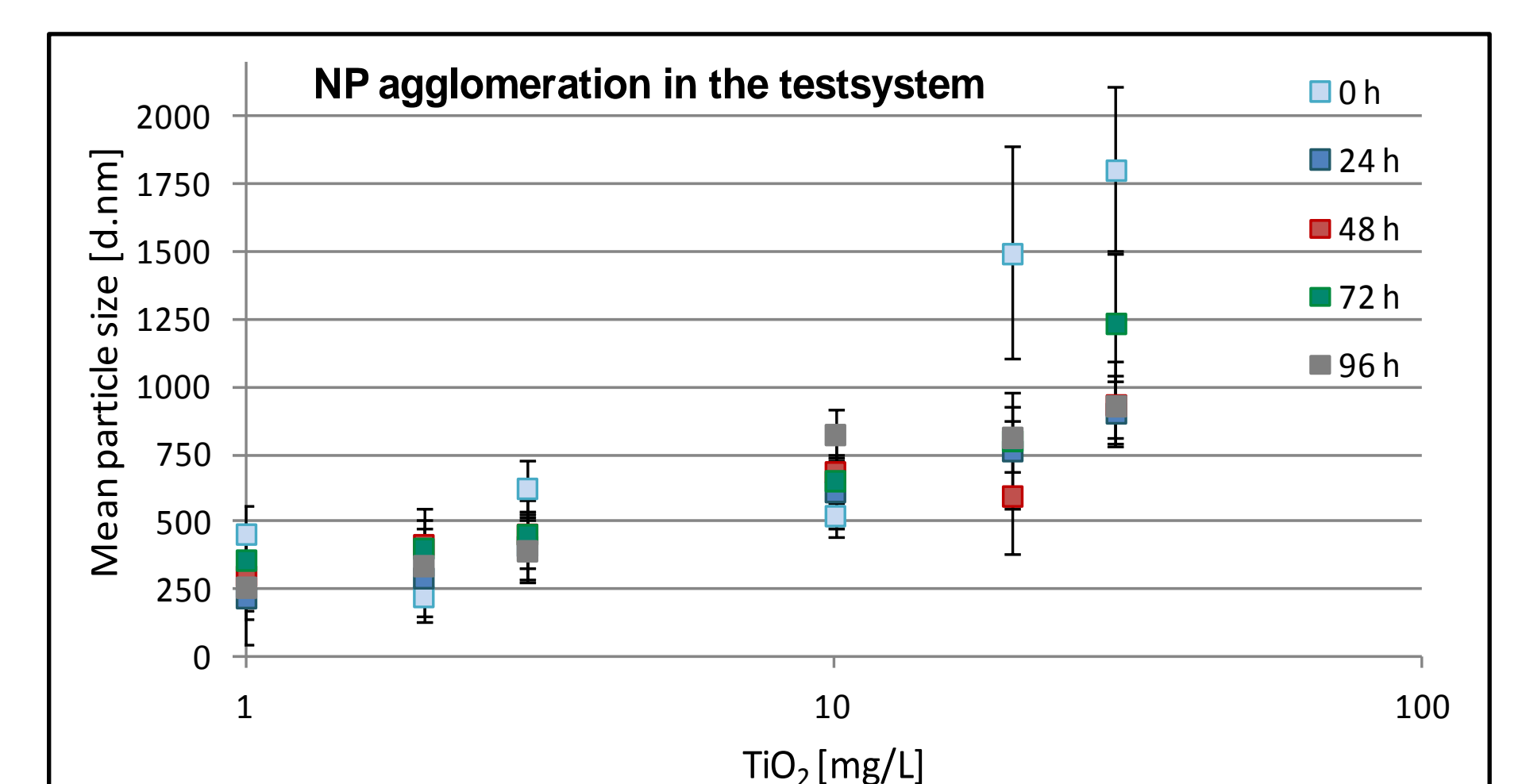


Fig. 5: DLS measures of TiO<sub>2</sub> dispersions introduced into the test media M9 after 0, 24, 48, 72 and 96 h. Average Intensity Mean values of particle diameter (10 repeated measures) and SD.

#### Scanning electron microscopy

Compared to DLS data:

- Particle sizes are of the same magnitude (453 nm +/- 58 nm for H<sub>2</sub>O suspensions; 413 nm +/- 77 nm for suspensions added to M9)
- Particle sizes do not increase with increasing concentrations
- Addition of M9 media has no impact on particle sizes

#### Acknowledgements

Thanks to Maximilia Kottwitz and Martina Nielsen for assistance in the lab, to Michael Gröger (ITMC UHH) and Jens Timmermann (TUHH) for help with DLS and SEM/EDX and to the OECD Sponsorship Programme for the provision of test material. Work was partially funded by Pro Exzellenza.

#### Reference

- [1] Gottschalk, F. et al., 2009. Modeled Environmental Concentrations of Engineered Nanomaterials (TiO<sub>2</sub>, ZnO, Ag, CNT, Fullerenes) for Different Regions. *Environmental Science & Technology*, 43(24), 9216-9222.
- [2] Dong et al. 2010. Photocatalytic degradation of phenanthrene and pyrene on soil surfaces in the presence of nanometer rutile TiO<sub>2</sub> under UV-irradiation. *Chemical Engineering Journal*, 158(3):378-383.